

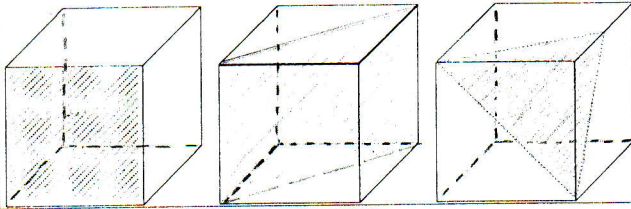
Answer the Following Questions:

(100 marks)

Question 1:

(25 marks)

- a) Solids are classified into crystalline and amorphous materials, study this statement, and describe the seven crystalline systems.
- b) Sketch the following planes in a cubic unit cell
a) $(0\ 3\ 2)$ b) $(4\ 4\ 2)$ c) $(\bar{1}\ \bar{1}\ \bar{2})$
- c) Determine the indices for the plans shown in following fig.



- d) Describe and explain how to use Miller indices to determine the spacing between crystalline planes.

Question 2:

(25 marks)

- a) Compare between face centred cubic (FCC), body centred cubic (BCC), and hexagonal close-packed (HCP) crystalline structures.
- b) The presence of attractive interatomic forces leads to the bonding in solids, describe and explain the several types of bonding.
- c) Write notes on the generation and absorption of x-rays and Bragg's law.
- d) Describe and explain the powder method to determine crystal structure by x-ray.

Question 3:

(25 marks)

- a) Explain the theoretical and real shear strengths of crystals.
- b) Derive an expression for the dependence of the heat capacity of solids at constant volume (C_V) on the temperature (T).

Question 4:

(25 marks)

- a) Write notes on
- a) relaxation time and mean free path.
- b) electrical conductivity of pure metals.
- b) In magnetic properties; explain the Hysteresis loop and the magnetic properties of solids.

With our best wishes